Code No.: CH102BS

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CMR ENGINEERING COLLEGE:: HYDERABAD UGC AUTONOMOUS

I-B.TECH-I-Semester End Examinations (Supply) - January- 2022 ENGINEERING CHEMISTRY (Common to CSM, ECE, MECH)

[Time: 3 Hours]

Note: This question paper contains two parts A and B.

[Max. Marks: 70]

Note.	Part A is compulsory which carries 20 marks. Answer all questions in Part A.	
	Part B consists of 5 Units. Answer any one full question from each unit. Each question	
	carries 10 marks and may have a, b, c as sub questions. PART-A (20 Marks)
1. a)	What is the disinfection of chlorination?	[2M]
b)	Explain Phosphate and Calgon conditioning.	[2M]
c)	Explain the rules of LCAO.	[2M]
d)	How n-type of semiconductivity is produced?	[2M]
e)	Write the reactions occurring at the electrodes in Daniel cell.	[2M]
f)	Define fuel cells and give applications of fuel cells.	[2M]
g)	Analyse the quality of coal by Proximate analysis.	[2M]
h)	Distinguish HCV and LCV.	[2M]
i)	Define Galvanising and Tinning.	[2M]
j)	Explain Vulcunisation of Rubber.	[2M]
37	PART-B (50 Marks)	
2.	a) Explain desalination of water by Reverse Osmosis.	[5M]
	b) Distinguish between Temporary and Permanent hardness. Explain various units to express hardness of water.	[5M]
	OR	
3.	Discuss ion exchange process for softening of hard water with a neat diagram.	[10M]
4.	 a) State the postulates of Molecular Orbital Theory. b) Explain molecular orbital energy level diagram of O₂ molecule with a neat diagram. 	[3M] [7M]
-	OR	[4M]
5.	a) Write salient features of crystal field theory.b) Discuss crystal field splitting of d-orbitals in octahedral geometry.	[6M]
6.	a) Explain construction and working of Calomel electrode.	[5M]
	b) Give a detailed account on the Lead-Acid cell with appropriate chemical reactions. OR	[5M]
7.	a) Explain the construction and working of H ₂ -O ₂ fuel cell.	[7M]
	b) What are batteries? Differentiate between primary and secondary cells.	[3M]
8.	Explain the process of refining of petroleum in detail with a neat diagram. OR	[10M]
9.	a) Differentiate Octane and Cetane Number.	[4M]
	b) Explain Fluid bed catalytic cracking with a neat diagram.	[6M]
10.		[3M]
10.	b) What are Biodegradable polymers? Explain preparation, properties and applications of Polylactic acid.	[7M]
	OR	100
11.	protection.	[6M] [4M]
	b) Write a note on Galvanic corrosion.	