

CMR ENGINEERING COLLEGE: : HYDERABAD
UGC AUTONOMOUS

I–B.TECH–I–Semester End Examinations (Supply) -June- 2025

ENGINEERING CHEMISTRY

(Common for CSD, CSM, ECE, MECH, AI&DS)

[Time: 3 Hours]

[Max. Marks: 70]

Note: This question paper contains two parts A and B.

Part A is compulsory which carries 20 marks. Answer all questions in Part A.

Part B consists of 5 Units. Answer any one full question from each unit. Each question carries 10 marks and may have a, b, c as sub questions.

PART-A

(20 Marks)

1. a) What are the Units of Hardness of Water? [2M]
- b) Define Osmosis and Reverse Osmosis. [2M]
- c) Define Atomic Orbital and Molecular Orbital. [2M]
- d) Discuss any two salient features of Crystal Field Theory. [2M]
- e) Derive Nernst's Equation. [2M]
- f) Differentiate Primary and Secondary cells. [2M]
- g) Define Octane number and Cetane number. [2M]
- h) Discuss any two applications of CNG and LPG. [2M]
- i) List out the types of corrosion. [2M]
- j) Discuss Functionality of monomers with an example. [2M]

PART-B

(50 Marks)

2. Explain the process involved in Ion-Exchange Process. [10M]
- OR**
3. Estimate Hardness of water by complexometric method. [10M]
 4. Discuss the Molecular Orbital Energy Level diagrams of O₂ and calculate the bond order and Magnetic nature of O₂. [10M]
- OR**
5. Explain the Crystal Field Splitting of transition metal ion d- orbitals in Tetrahedral complexes. [10M]
 6. Discuss the reactions involved in the Pb-Acid storage battery with a neat diagram. [10M]
- OR**
7. Construct the Calomel electrode with a neat diagram and explain the reactions involved in it. [10M]
 8. Write a brief note on Proximate and Ultimate analysis of Coal and its significance. [10M]
- OR**
9. Determine the calorific value of Junker's gas calorimeter. [10M]
 10. Explain the mechanism involved in Electrochemical corrosion of rusting of Iron. [10M]
- OR**
11. Discuss the Chain growth polymerisation mechanism with suitable example. [10M]
