Code No.: R22CH202BS

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H.T.No.

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CMR ENGINEERING COLLEGE: : HYDERABAD UGC AUTONOMOUS

I-B.TECH-II-Semester End Examinations Regular) - June- 2025 ENGINEERING CHEMISTRY

(Common for ECE, CSD, CSM, IT, CSC)

(Common for ECE, CSD, CSW, 11, CSC)		
[Time	e: 3 Hours] [Max. Mar	·ks: 60]
Note: This question paper contains two parts A and B.		
Part A is compulsory which carries 10 marks. Answer all questions in Part A.		
Part B consists of 5 Units. Answer any one full question from each unit. Each question		
carries 10 marks and may have a, b, c as sub questions.		
DADT A (10 M 1)		
	$\underline{\mathbf{PART-A}} \tag{1}$	0 Marks)
1. a)	What is Calgon conditioning?	[1M]
b)	Define hardness water.	[1M]
c)	Relate between HCV and LCV of a fuel.	[1M]
d)	What is the significance of Cetane number?	[1M]
e)	Illustrate applications (any two) of electrochemical series.	[1M]
f)	What is a battery and write its classification.	[1M]
g)	Explain pitting corrosion of metal.	[1M]
h)	Define corrosion of metals.	
		[1M]
i)	Explain why natural rubber needs vulcanization.	[1M]
j)	Summarize various applications biodegradable polymers.	[1M]
	PART-B (5	50 Marks)
2.	Explain the purification of water by Ion exchange process and give its advantages.	[10M]
۷.	OR	[TOIVI]
3.	Summarize a short note on the following	[10M]
Э.	i) Phosphate conditioning ii) Caustic embrittlement iii) Ozanization	[TOIVI]
	1) I nospitate conditioning it) Caustic emorittement iti) Ozanization	
4.	What is cracking? Discuss the Moving bed catalytic cracking method to obtain	n [10M]
	gasoline from heavy oils.	F . J
	OR	
5.	Discuss the manufacture of gasoline by Fisher-Tropsch method.	[10M]
		[]
6.	Explain the construction and working of Lead acid battery. Write down the reactions	s [10M]
	taking place during charging and discharging.	[]
	OR	
7.	Explain the construction of Quinhydrone electrode and write determination of p ¹	^H [10M]
<i>,</i> .	using Quinhydrone electrode.	[10141]
	using Quinity at one electrode.	
8.	Write a short note on the following	[10M]
0.	i) Sacrificial anodic protection method ii) Galvanizing	[TOIVI]
	OR	
9.		[10 M]
9.	Explain the Electrochemical theory of wet corrosion, giving its mechanism.	[10M]
10.	Discuss the preparation, properties and applications of Bakelite and Teflon.	[10M]
10.	OR	
11.	Analyze the doping mechanism of conducting polymer (p-doping & n-doping).	[10M]
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