

CMR ENGINEERING COLLEGE: : HYDERABAD
UGC AUTONOMOUS

I-B.TECH-I-Semester End Examinations (Regular) - December - 2025
ENGINEERING CHEMISTRY
(CSE)

[Time: 3 Hours]

[Max. Marks: 60]

Note: This question paper contains two parts A and B.

Part A is compulsory which carries 10 marks. Answer all questions in Part A.

Part B consists of 5 Units. Answer any one full question from each unit. Each question carries 10 marks and may have a, b, c as sub questions.

PART-A

(10 Marks)

1. a) Define hardness of water and its classification. [2M]
- b) List any four Characteristics of a good fuel. [2M]
- c) What is Cathodic protection? Name two methods used for it. [2M]
- d) Write any two applications of Conducting polymers. [2M]
- e) Write any two applications of Quartz in engineering. [2M]

PART-B

(50 Marks)

2. Explain the Ion-exchange process for softening of water. Compare it with Reverse Osmosis for desalination of brackish water. [10M]
- OR**
3. Explain the estimation of hardness of water by Complexometric (EDTA) titration [10M] method.
 4. Explain the construction, working, and applications of a Direct Methanol Fuel Cell [10M] (DMFC) with a neat labeled diagram. Discuss its advantages and limitations.

OR

5. Explain Fischer tropsch's process for synthesis of petrol with neat diagram. [10M]
6. Explain the Electrochemical theory of corrosion with a neat diagram. [10M]

OR

7. Describe the construction, working, and applications of the Calomel electrode. How is it used as a reference electrode in the determination of pH of an unknown solution? [10M]

8. Explain the preparation and applications of PVC, Nylon-6, 6 and Buna-S rubber. [10M]

OR

9. Explain the classification of Conducting polymers. Describe the mechanism of conduction in trans-polyacetylene with a neat diagram. [10M]

10. Describe UV-Visible spectroscopy and its use in the analysis of pollutants in the dye industry. [10M]

OR

11. Explain the classification of Smart Materials with suitable examples. [10M]
